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Bju chemistry chapter 8 review answers

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A burner & fuel is also required & not included in the kit; we recommend purchasing a Lab Burner, but an Alcohol Burner can also work! An economical alternative to the BJU Press chemistry labs is our MicroChem Kit. This BJU chemistry lab kit includes 17 labs that correlate with 11 chapters of the BJU Press Chemistry text. While this is a less rigorous lab experience, it is more cost-effective, does not require a balance or burner, and still provides your high schooler with a well-rounded chemistry lab experience. This high school science lab kit is for the 4th Edition curriculum. If you need materials for an earlier edition, please email our wonderful Customer Service team and they will happily help you find everything you need. Please Note: This BJU chemistry lab kit does not coincide with BJU Press Distance Learning curriculum. Click here for information on how to order the materials needed for the Distance Learning labs. 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Acetic acid, 30 ml Acetone, 30 ml Alligator clip leads, 2/pack Ammonium chloride, 30 g Bar magnet, 1" ceramic, 2/pack Battery, 9-volt, alkaline Beaker, glass, 150 ml, HST Beaker, glass, 250 ml, HST Beaker, glass, 50 ml, HST Beaker, glass, 600 ml, HST Capillary tubes, vial of 100 Carbon, electrode red, 1/4" x 4" Clay, modeling, 1/4 lb. Copper II (cupric) sulfate, 30 g Copper metal shot, 30 g Copper, powder, 30 g Cork, size 9 Crucible & lid, 15 ml, high form Ethyl alcohol, 95%, 30 ml Evaporating dish, 75 ml Filter paper, 11 cm, 10 sheets Flask, Erlenmeyer, 250 ml Funnel, glass, 75mm diameter Grad. cylinder, glass, 10 ml Grad. cylinder, glass, 100 ml Hydrochloric acid, 30 ml Hydrogen peroxide, 30 ml Iron II (ferrous) sulfate, 30 g Iron III (ferric) chloride, 30 g Iron metal filings, 30 g Isopropyl alcohol, 90%, 30 ml Lab apron, white vinyl Lithium chloride, 30 g Magnesium ribbon, 60 cm Manganese dioxide, 15 g Marbles, pack of 15 (sold only in kits) Methyl alcohol, anhydrous, 30 ml Methyl orange, 0.1% solution, 30 ml Methyl red, 0.02% solution, 30 ml Mineral oil, Klearol, 30 ml Mortar and pestle, 80 ml Pinch clamp Pipet (medicine dropper) Potassium chloride, 30 g Potassium nitrate, 30 g Potassium permanganate, 30 g Potassium thiocyanate, 30 g Red LED, 5mm diameter Resistor, 1K ohm, 1/4 watt Ring support stand, 4"x6" base, 18" rod Ring support, 4" diameter Rubber stopper, #6.5, 2-hole Rubber stopper, #0, 1-hole Rubber stopper, #0, solid Safety gloves, size 8-8.5 medium Safety goggles, full-size, splash style Salicylic acid, powder, 30 g Sand, fine, white, 1 lb. Sandpaper, 1 sheet medium grit Sodium bicarbonate, 30g Sodium hydroxide, 30g Sodium nitrate, 30 g Spatula, stainless steel, 15cm Spectroscope, Quantitative Stirring rod, glass, 10" long Strontium chloride, 10 g Styrofoam Cup Lid, each Styrofoam Cup, 8 oz, each Sulfur, 30 g (flammable solid) Sulfuric acid, 30 ml Test tube clamp Test tube rack, 20mm, 40 holes Test tubes, extra large 20x150 mm Test tubes, small 13x100 mm Thermometer, Celsius, 12" Thymol blue, 15 ml Tongs, crucible Triangle, clay, 2" pipe-stems Tube, 5 mm glass, 12" long Tube, 2 pack, 5 mm glass, 3" long Tubing, 4.8 mm rubber, 2' long Vernier Caliper Wash bottle, 250 ml Watch glass, 100 mm diameter Wax pencil (marks glass) - Red or Black Wire gauze, ceramic center, 5" square Wire, copper, 14 gauge, 6' long (sold only in kits) Wire, insulated copper, #24, 15 ft. Wooden splints, 20 pack Zinc, metal electrode, 4" SPECIFICATIONS TAB BJ-KIT11 Shipping Restriction: This item only ships Economy or UPS to a street address in the 50 US states. Warning: HAZARDOUS CHEMICAL - This kit contains hazardous chemicals. Not for children under the age of 15. For use under adult supervision. Read warning labels carefully. Homeschool Curriculum & Kits/BJU Press University Press/Science Grade 11 Chemistry/science-curriculum-kits/./science-curriculum-kits/bob-jones-university-press/./science-curriculum-kits/bob-jones-university-press/science-11-chemistry/ Questions & Answers Chemistry (4th edition) focuses on the composition and structure of atoms. Students will learn about chemical structures, bonding, the names of substances and their formulas, and chemical equations. Most importantly, Mr. Bill Harmon shows the level of order and sophistication that went into creating the universe. 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